Electrical Properties Of The OLED With Convergence Relation By Surface Treatment Methods

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Abstract: ITO (indium-tin-oxide) is used widely as the anode electrode in organic electroluminescence devices (OLEDs). Several studies have dealt with the effect of ITO chemical treatment but to the OLEDs performance have not been considered the effect of this treatment. We were present here results regarding these effect. We studies of the effect of surface treatments on the properties of ITO substrate. We were performed chemical treatment with RCA and Aqua regia of the OELDs. Since we will understand the properties of OELDs, we were performed Fowler-Nordheim tunneling. The buffer layer and ITO surface treatment were performed to increase the efficiency of the OLEDs. As a result, we were obtained confirm the surface properties of ITO can be significantly changed by chemical treatment. The oxygen decrease produced by chemical treatment lead to increase of oxygen vacancies. We confirmed that the oxygen vacancies role on increasing carrier injection. A study on the cause of improving the performance of the device using the surface treatment and the usingthe buffer layer will be of great help in understanding the relationship between the properties of the ITO substrate and the organic electroluminescent device.

Keywords: Indium-tin-oxide (ITO), OLED, Fowler-Nordheim tunneling, RCA treatment, Aqua regia treatment.

1. Introduction

The improvement in luminous efficiency can be due to the possibility of manufacturing an amorphous organic thin film having good characteristics and thus improving the injection of electrons or holes. In the early development stage, the development of various luminous colors was the main focus, and studies on white luminescence [1-4] were actively conducted as an extension of these. Efforts to improve luminous efficiency were attempted at the early stage of the doping method, and then efforts were made through the development of high-lumance materials. As an understanding of the operating principle of an organic electroluminescence device (OLED), a high-lumance light-emitting material using the same has been developed. However, since the development of light-emitting materials is slow and requires a lot of initial investment, researchers first understand the driving principle of the device, and from this, focus on finding necessary elements for improving device efficiency, and these efforts are classified as follows [5-8].

In order to improve the efficiency of an OLED, 1) improvement of transporter injection, 2) a structure of a device that can efficiently recombine with a numerical balance of electrons and holes, and 3) development of materials with high luminous efficiency are required [9, 10].

In this paper, among the various surface treatment methods of ITO substrates, the surface of the ITO substrate is stabilized through a chemical treatment and the efficiency of the OLED is improved by adjusting the surface characteristics suitable for organic electroluminescent devices. It was intended to investigate the phenomenon at the interface between the substrate and the organic material. RCA and Aqua regia solutions were used as a chemical treatment method to examine the changes in the characteristics of the device when the surface and composition of the ITO substrate was changed, and investigate how the surface changes affect the characteristics of the device.

2. Materials

As the material used in this experiment, Alq3, a metal complex compound, was used as a light emitting layer of an organic electroluminescent device. And diamine-based TPD was used as the hole transport layer. The molecular structure of TPD and Alq3 are shown in Figure 1. Alq3 is a chelate compound surrounded by an 8-hydroxyquinoline ligand with a metal ion, Al as the center, and is a metal complex that has been studied since it was first known by Kodak. Alq3 is a green fluorescent material, has very high quantum efficiency, and has good luminescence properties as a light-emitting material because it is a very stable film formation, good transporter movement, and a good thermal resistor. TPD is a material having the most excellent properties among materials used as a hole transport layer, and the reason for using TPD and Alq3 is that these two materials are the most basic materials used in organic electroluminescent devices and exhibit good properties.

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3. Results

3.1 C-V (Cyclic-Voltammetry) Characteristics

Cyclic Voltammetry was used to measure the energy gap of the organic material in this experiment. Figures 2 show the results of measuring the cyclic voltammetry of the Alq₃ and TPD used in the experiment. By knowing the energy gap of a material by cyclic voltammetry, it is possible to confirm the process by which the carrier can inject more efficiently and increase the luminous efficiency. By measuring the voltage at which oxidation and reduction of a material begins by circulating voltage current and adding 4.87 eV which is the correction value of the cyclic voltammetry for the UPS, we can find the ionization energy and electron affinity of the material. From the result of measuring the cyclic voltammetry, it can be seen that in the case of TPD, the oxidation reaction is +0.67 eV and the reduction reaction is -2.30 eV. Therefore, adding the correction value of 4.8 eV results in ionization energy of 5.47 eV and an electron affinity of 2.50 eV. As a result, it can be confirmed that the energy gap of TPD is 2.97 eV, which is the difference between ionization energy and electron affinity. In the same way, in the case of Alq₃, the oxidation reaction is +0.98 eV, the reduction reaction is -1.66 eV, the ionization energy is 5.78 eV, the electron affinity is 3.14 eV, and the energy gap is 2.64 eV.

![Figure 1: Molecular structure](image)

(a) Alq₃, (b) TPD

3.2 XPS and Binding Energy spectra of the ITO

Figure 3 shows the results of XPS (X-ray photoelectron spectroscopy) analysis of the ITO. In Figure 3, the constituents of ITO without surface treatment and the mostly indium (In), tin (Sn), and oxygen (O), and the surface contains carbon (C) due to residual organic matter. In the case of indium, the spectrum appears at 444.8 eV and the spectrum appears at 452.4 eV. Each spectrum represents a spectrum of photoelectrons from In 3d₃/₂ and In 3d₅/₂. Looking at the spectrum of the In 3d at the surface and bulk of ITO, respectively, there is no change in the binding energy between the surface and the bulk. The spectrum due to the photoelectrons of Sn 3d appears from the tin. At 486 eV, the spectrum due to Sn 3d₃/₂ and the spectrum in the binding energy region due to electrons of Sn 3d₅/₂ are observed. Like indium, tin does not show any change in the spectrum of the surface and bulk. The absence of these spectral changes indicates that no structural defects or new bonds are created. The component that shows the greatest change in binding energy in ITO is oxygen. In the case of oxygen, a change in the spectrum between the bulk and the surface appears due to a new bond with external hydrogen. The oxygen spectrum appears in the binding energy region corresponding to the photoelectron of O 1s. As shown in Figure 4, the surface shows a change in spectrum due to the influence of the new bond. As shown in Figure 4, there are three types of oxygen present on the surface. The first (O₁) is the binding energy by O₂⁻ (530.2 eV), which means the binding energy by oxygen that is perfectly bonded. The second (O₂) is the binding energy by
$O_2^-$ (531.7 eV), which means the binding energy by photoelectrons of oxygen adjacent to the oxygen vacancy. Finally, the third ($O_{III}$) means the binding energy (533.4 eV) by oxygen (-OH) bound to the surface hydrogen.

![In 3d core level spectra](image1)

![Sn 3d core level spectra](image2)

![O 1s core level spectra](image3)

**Figure 3** XPS spectra of the ITO substrate.

**Figure 4** Binding energy of oxygen species of the ITO surface.

### 3.3 Chemical treatment (RCA, Aqua regia)

There are two main types of chemical treatment methods. The first is a method of treating the ITO substrate by mixing aqueous ammonia, hydrogen peroxide, and secondary distilled water, and this method is called the RCA treatment method. The second method is to mix nitric acid, hydrochloric acid and secondary distilled water for treatment. The method of treating the surface with an acid is called Aqua regia treatment.

In the RCA solution, ammonia water (NH$_4$OH), hydrogen peroxide (H$_2$O$_2$), and secondary distilled water were mixed in a ratio of 1:4:25, and the ITO substrate was put in a state where the temperature was heated to 60°C. The Aqua regia solution was prepared by mixing nitric acid (HNO$_3$), hydrochloric acid (HCl), and secondary distilled water at a mixing ratio of 1:3:25, put an ITO substrate, treated the surface, and cleaned the device. In addition, the ITO substrate surface was treated alternately between the RCA treatment with the best device performance and the Aqua regia treatment with the best device performance. For the improvement of device performance by chemical treatment, the electrical characteristics are investigated through current density-voltage, luminance-voltage, and Fowler-Nordheim tunneling, and surface characteristics are analyzed for the main cause that affects the actual device performance through the XPS.

### 3.3.1 RCA treatment
Figure 5 shows the voltage-current density curve of the device when the surface is not treated with RCA and when the surface is treated. It can be seen that when the RCA treatment was treated for 5 min and 10 min, the carrier injection characteristics were improved by 10 times compared to the case without treatment. It can be seen that the emission intensity of the device is further improved when the processing time is 5 and 10 min, compared to the case of not processing.

In the case of 5 min of RCA treatment at 10V, the current density is shown as a 20 mA/cm$^2$. Compared with the case of 10 min treatment, it could not be seen a little difference in the current density. However, it can be seen that there is a difference in luminance between the 5 min treatment and the 10 min treatment in the luminance characteristics. At 10V, the 5 min treatment is about 200 cd/m$^2$ and the 10 min treatment is 100 cd/m$^2$.

The difference in luminance occurs at almost the same current density, indicating that the RCA 5 min treatment improved the efficiency. As a result, it was confirmed that the most efficient condition in the RCA treatment was 5 min treatment.

It can be seen that the cause of the device performance improvement according to the RCA process is due to the increase of the injected carriers as confirmed through the current density-voltage and luminance-voltage characteristics. To confirm this increase in transporter injection, it can be analyzed by applying the Fowler-Nordheim tunneling theory. Figure 6 shows a Fowler-Nordheim plot was plotted using the current density-voltage curve obtained from the actual device.

When the ITO surface was chemically treated, the spectrum of the surface and bulk of indium, tin, and oxygen is shown in Figure 7, 8, 8. It was confirmed that the surface and bulk of indium and tin did not change as shown in the spectrum. The biggest change that is consistent is the change in oxygen on the surface. In the case of chemical treatment as oxygen, the element that affects the electrical conductivity in the ITO substrate, it can be seen that the change in other composition is inconsistent.
However, it can be seen that the change in the composition of oxygen is consistently changed depending on the treatment conditions. When comparing the performance of the device according to the change in the oxygen composition on the surface, it can be seen that the performance of the device is improved as the composition of oxygen decreases. Also, it was confirmed through device characteristics that the amount of holes injected increases according to the composition ratio of oxygen, and that the increased amount of holes contributes to light emission of the device.

3.3.2 Aqua regia treatment

Figure 10 shows the current density-voltage characteristics of the device, which appears when the device is directly manufactured by treating the surface of the ITO substrate. In the case of Aqua regia treatment, it can be confirmed through the current density-voltage characteristics that the injection of the carrier is very higher than that of the untreated device. Depending on the treatment time, in the case of 5 min and the injection of carriers into the device increased the most, followed by 10 min. In the current density-voltage characteristics, the luminance-voltage characteristics also show that the amount of transporters contributing to direct light emission increases in proportion to the amount of carrier injected in the order of 5 min and 10 min depending on the processing time. We were obtained the current density value when treated for 5 min of Aqua regia at 10V is 60
mA/cm², and 30 mA/cm² when treated for 10 min. When the treatment time was 5 min, it was found that the current density increased by about 2 times. The luminance characteristic is 1000 cd/m² when 5 min treatment is performed at 10V, and 300 cd/m² when 10 min treatment is performed. It was confirmed that the luminance characteristics also increased with the increase of the injected transporter.

This is due to the variation of the effective potential barrier height obtained from the Fowler-Nordheim plot of the current density-voltage characteristic curve. Like the RCA treatment, the current density-voltage characteristics of the device treated with Aqua regia are also shown in Figure 11 and we were observed that it is well applied to the Fowler-Nordheim tunneling theory. When the Aqua regia treatment was performed rather than the RCA treatment, the operating voltage was slightly decreased. However, it was found that the operating voltage did not change according to the conditions of Aqua regia treatment.

![Current density-voltage and luminance-voltage plots](image)

(a) Current density-voltage  
(b) luminance-voltage

**Figure 10** Electrical Properties with ITO substrates of Aqua regia treatment.

![Fowler-Nordheim plots](image)

(a) Aqua regia 5min  
(b) Aqua regia 10min

**Figure 11** Fowler-Nordheim plot of the OLED with Aqua regia treatment.

4. Conclusions

Since the surface treatment of the ITO substrate is essential to improve the performance of the organic electroluminescent device, we tried to investigate the effect and cause of the surface characteristics of the ITO substrate on the performance of the device. The surface treatment of the ITO substrate was chemically treated (RCA, Aqua regia) to analyze the characteristics of the device.

The improvement of device performance by chemical treatment could result in an increase in transporter injection due to a decrease in the effective potential barrier. This phenomena was occurs when carrier flows at the interface between the ITO substrate and the organic material. In the case of RCA treatment, the height of the effective potential barrier decreased by 0.08 eV, and even in the case of Aqua regia treatment, 0.08 eV was decreased. Also, in the case of mixing treatment, it was found that the most decrease was 0.14 eV. The carrier injection increases as the height of the effective potential barrier decreases, and the increase of the injected carrier improves the device performance. As for the composition change of the surface of the ITO substrate by chemical treatment, oxygen changed the most. It can be seen that the RCA treatment decreased by 5%, the Aqua regia treatment decreased by 6%, and the mixed treatment decreased by 11%.

Also, it was confirmed that the O₁ component, which is the factor that most affects the device performance, decreased to 7% in the case of RCA treatment, 12% in the case of Aqua regia treatment, and 20% in the case of mixed treatment. The decrease in the O₁ component is the main reason for increasing the oxygen vacancy, and the increase in the oxygen vacancy increases the number of carriers. It can be concluded that an increase of carriers injected into the organic material consequently improves the device performance.
7. References