

Solidification, Microstructure, and Mechanical Properties of Mg-Nd-Gd-Zn-Zr Magnesium Alloy with 1.5 Samarium

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Abstract: The influence of samarium (Sm) content on the solidification characteristics, microstructure, and mechanical properties of Mg-2.8wt%Nd-1.5wt%Gd-0.5wt%Zn-0.5wt%Zr magnesium alloy was studied. The cooling curves and microstructure analysis results showed that Sm rare earth element refined the grains of the alloys, where the solidification time of α -Mg phase of decreased as addition of Sm, which reflected to the microstructure of alloy and the grains became refined, also Sm combined with the initial phase of the intermetallic base alloy and crystallized along the grain boundaries. In addition, Mg₄₁Sm₅, Mg₃Zn₆Sm and (Mg, Zn)₃Sm new intermetallic phases were formed as addition of Sm. Both grain refinement and formed intermetallic phases led to the improvement in hardness and tensile strength.

Keywords: Magnesium alloys, rare earth, thermal analysis, microstructure, mechanical properties

1. Introduction

Reduced weight of products can be attained by many approaches such as redesign, removal of unnecessary parts, reducing thickness and selection of light materials. For such reasons, much effort has been spent on developing lighter materials, such as composites, and Al and Mg alloys for vehicles, and aerospace industrials (Ferro, Saccone et al. 2013). Magnesium is the 6th most plentiful component in the world's covering and is the lightest of every single auxiliary metal with a high specific stiffness (Ali, Qiu et al. 2015). The focus on Mg-RE (RE such as Sm, Yt, Pr, Gd, etc.) alloys is attributed to the significant strength at the normal and relatively high temperatures, and also because of their practical creep resistance (Easton, Gibson et al. 2018). Recent studies highlighted that some rare earth (RE) metals, such as Yttrium (Y), Neodymium (Nd) and Samarium (Sm) are very effective elements to improve the mechanical properties of Mg alloys. Mechanical properties of magnesium can be enhanced by adding different contents of element-alloy which results improvements in increased tensile strength, creep resistance, thermal stability, or corrosion resistance. Thermal analysis techniques are performed to investigate the solidification characteristics of metals and their alloys. Recently, in industrial applications, computer-aided cooling curve analysis (CA-CCTA) is common method, and has been successfully applied to several investigations on the solidification sequences of magnesium alloys (Farahany, Bakhsheshi-Rad et al. 2012), (Lotfabadi, Idris et al. 2013). In this paper, the solidification behavior, microstructure, and mechanical properties of Mg-Nd-Gd-Zn-Zr magnesium alloy with samarium additions are investigated.

2. Experimental

In this paper samarium (Sm) rare earth element with (1.5wt.%) added into the Mg-2.8wt%Nd-1.5wt%Gd-0.5wt% Zn-0.5 wt% Zr magnesium alloy. The base alloy was melted at a temperature of 750 °C in an induction furnace using Ar and 2% of SF₆ mixed gas. The solidification characteristics of cooling curves of the investigated alloys were determined via CA-CCTA, where two K-type thermocouples are used, one of them was placed in the center of the mould and the other one is closed to the wall. The solidified samples were cut from the central section of ingots and then prepared according to the standard of ASTM E3 to be used for metallographic examination. The optical microscope (OM) observed the microstructure of the alloy, and a scanning electron microscope (SEM). The samples of tensile test were produced using the permanent casting process with steel mould, and each casting provided three tensile bars. The bars of final test were cut and machined using a CNC machine according to the standard dimensions of ASTM B557, and the reported values are the average of at least three measurements. Hardness test is performed using Vickers hardness (Hv) on the same samples that are used for microstructure analysis according to the standard of ASTM E92, with the 5 Kg, applied load and 10-second load holding time, and the reported values of hardness are the average of seven readings for any sample.

3. Results and Discussion

3.1 Solidification Characteristics

Fig. 1 shows the cooling curves that extracted during solidification process of the investigated alloys.

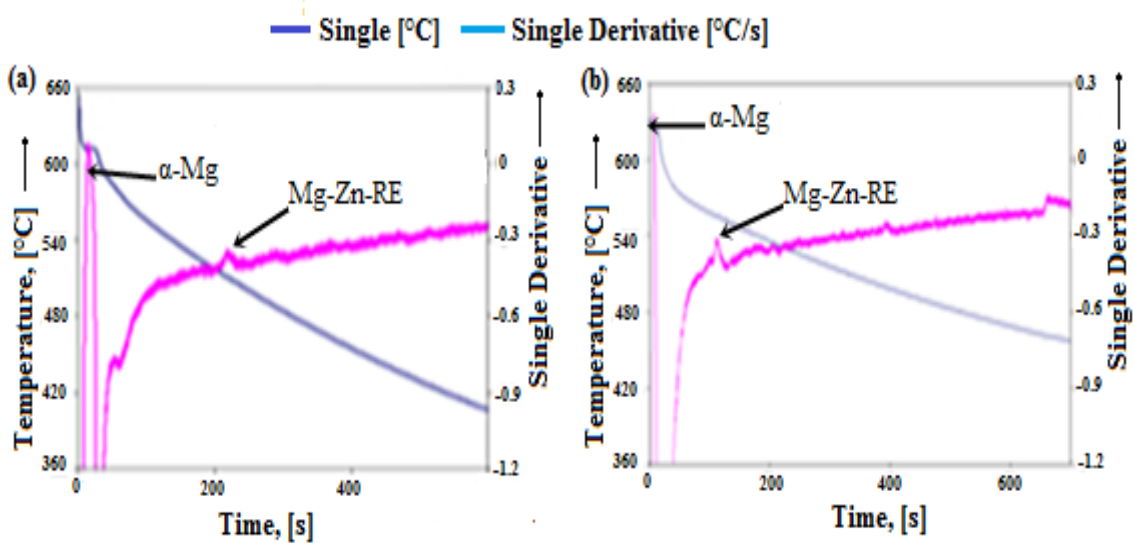


Figure 1. The cooling curves of. (a) Base alloy (b) Treated with 1.5 wt. % Sm

The nucleation and growth of the primary α -Mg phase are the first stage of solidification process of Mg alloys, which followed by other formed phases containing the alloying elements. From the first derivative of cooling curves, the main characteristic parameters of α -Mg phase have been determined, including nucleation temperature ($T_{N\alpha-Mg}$), and growth temperature ($T_{G\alpha-Mg}$) of α -Mg phase. The start of liquid peak (on the first derivative curve) is denoted by temperature ($T_{N\alpha-Mg}$), which is considered as the beginning of the solidification (Song, Wu et al. 2014).

At 635.9 °C, α -Mg phase of base alloy is formed ($T_{G\alpha-Mg}$), and as temperature reached 645.8 °C, the solidification begins ($T_{N\alpha-Mg}$). The solidification temperature started in Mg-10Gd-3Y-0.4Zr alloy at 640.2°C (Song, Wu et al. 2015), and α -Mg phases of the as-cast Mg-4Sm-4Zn-0.5Zr alloy formed at 642 °C (Che, Cai et al. 2017). The primary stable dendrites start to solidify from the melt at nucleation temperature ($T_{N\alpha-Mg}$), which decreased to 641.3 °C as 1.5 wt. % of Sm added due to forming of third phase which detected via XRD. The above results showed that a part of Sm dissolves into α -Mg matrix during the solidification of the alloy. Moreover, high melting point compounds $Mg_{41}Sm_5$ act as a bar to the growth of α -Mg grain, which lead to decreasing the nucleation temperature of α -Mg phase (Zhou, Li et al. 2015). The solidus temperature decrements as result of decrease the nucleation temperature of α -Mg phase and simultaneously widened the solidification range (Dobrzański, Król et al. 2010). Addition of Sm, caused to segregate the second phase gradually, and reduce the nucleation particles of α -Mg (Chen, Jin et al. 2015). Furthermore, forming a new intermetallic compound (Mg_{3Zn6Sm} ternary phase) which detected by XRD at the grain boundaries of microstructure of treated alloy that led to decrease of the nucleation and growth temperature of primary magnesium dendrite. Addition of Sm results in it enrichment into front of the solidification interface, then the enrichment Sm effectively prevents the other alloy elements from diffusing to the matrix α -Mg, which led to significant increase in quantity of the eutectic phases (Zhang, Huang et al. 2017).

As addition of Sm, the solidification time α -Mg phase ($\Delta t_{\alpha-Mg} = t_{G\alpha-Mg} - t_{N\alpha-Mg}$) decreased, which leads to grain refinement of the alloys. It has been reported that addition of rare earth (RE) into Mg-Zn alloys had an important effect on changing the solid solubility of the Zn in the matrix as a result of decrement in the solidus curve, which shortened the time of nucleation and reduced the grain size (Li, Du et al. 2012, Singh, Srinivasan et al. 2014). Mg-Zn-RE intermetallic compound formation was the last stage of solidification process, and the growth temperature ($T_{GMg-Zn-RE}$) and growth time ($t_{GMg-Zn-RE}$) have been determined the temperature and time of end of solidification (solidus temperature and time), (T_s & t_s) were found too. As a result, the growth temperature of Mg-Zn-RE phase of base alloy was 526.8 °C, which decreased to 526 °C with addition of Sm. The results of solidus temperature (temperature of solidification end) showed almost similar behavior with growth temperatures of Mg-Zn-RE phase, where obtained at 521.7 °C with addition of Sm compared with 521.4 °C for the alloy without treated. The solidification temperature (solidification range, $\Delta T = T_{N\alpha-Mg} - T_s$), and frozen time (solidification time, $\Delta t = t_s - t_{N\alpha-Mg}$) of the alloys have been obtained as well. The frozen time of base

alloy was obtained as 220.56 s, which decreased significantly to 173.19 s with addition of 1.5 wt. % of Sm. Mg and Sm have almost similar atomic radius (Magnesium atom-0.16 nm, Samarium atom-0.179 nm), and according to the metallographic theory, Sm element usually acts as heterogeneous crystalline core which lead to the refinement of grains, and shortens the nucleation (Xu, Feng et al. 2017). The solidification temperatures results showed a high change, which receded from 124.8 to 119.6 °C as the base alloy treated with Sm. The solidification temperature clearly decreased and diffusion of Sm atoms become harder, nucleation of new phase formed at low temperature thereby widen the solidification temperature and solidification time in the present investigated alloy.

3.2 Microstructure Analysis

Optical micrographs are shown in Figure 2. As a result, the microstructures of alloys consist of α -Mg matrix and second phase crystallized along the grain boundaries, which showed a kind of massive morphology with dark contrast. In addition, microstructure of base alloy appears to have a greater density of fine second phase particles within the grain as well as a few coarse precipitates at the grain boundaries.

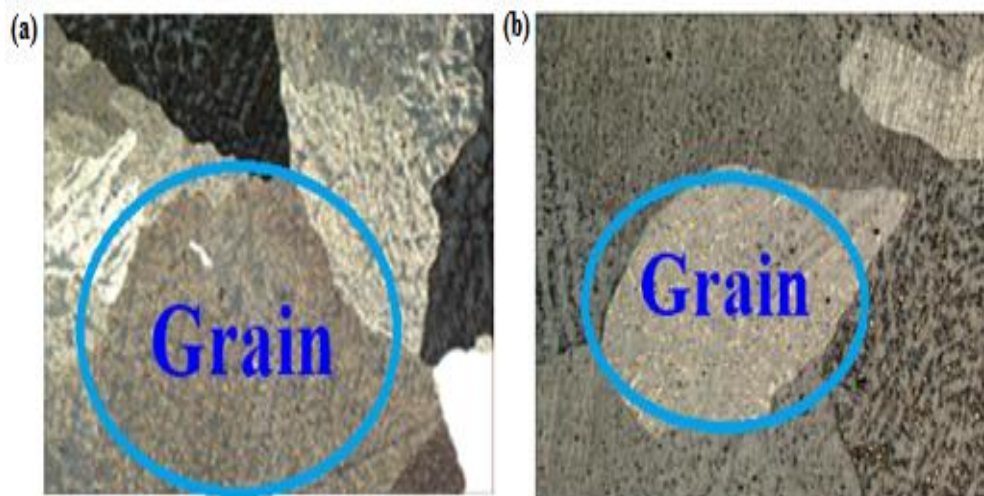


Figure 2. Microstructures image of: (a) base alloy and (b) base alloy treated with 1.5 wt. % (x50)

The grain sizes of alloys (0 and 1.5) wt. % Sm are 383.84 μm and 223.66 μm , respectively, this result showed that Sm addition demonstrate to be beneficial to refining the grain size of investigated alloy. The reduction in grain size occurred due to the reduced solidification time ($\Delta t_{\alpha\text{-Mg}}$), and heterogeneous nucleate particles, where Sm rare earth element restricted the growth of grains (ZHANG, HUANG et al. 2017). It has been reported that the grains of Mg alloys microstructure refined due to heterogeneous nucleation of α -Mg primary phase on spherical particles and suppressed grain growing process (Guan, Meng et al. 2019).

The volume fraction of the intermetallic Mg-Zn-RE phase increased obviously as addition of Sm, which obtained 28.67 % and 35.16 % for base alloy and alloy treated with 1.5 wt. % of Sm respectively. Results of previous studies showed that the volume fraction of intermetallic phases increased as increase Sm content. Also, the microstructure obtained from adding Sm into Mg alloys mainly consists of α -Mg matrix and second phase (such as Mg₄₁Sm₅) phase witch distributed at the grain boundaries, and the amount of the second phase increased with increase Sm content. (Sun, Hu et al. 2015, Zhang, Huang et al. 2017, Guan, Li et al. 2018).

As shown from SEM micrographs, the microstructure of untreated alloy consists of α -Mg grains marked A, surrounded by second phase crystallized along the grain boundaries as a kind of massive morphology (Figure 3-a) marked B. Moreover, as a result of EDS spectrum for regions A and B, the intermetallic secondary phase was composed of magnesium, zinc, gadolinium and neodymium, (Marked B) and the matrix contains magnesium and a small amount of zirconium, (Marked A). SEM micrographs of the alloy treated with Sm showed that its microstructure was composed of primary α -Mg phase and intermetallic eutectic phase distributed along grain boundaries as well as a few isolated particles inside the grains also a third phase coexisted with the secondary phase. Two different regions and morphology for intermetallic compounds, which crystallized along the grain boundaries; light grey region (Figure 3-b marked D) and dark grey region (Figure 3-b marked C). The light grey region consists of Mg-Zn-Nd-Sm, and the dark grey phase consists of Mg-Nd-Gd-Zn-Sm, which formed as a third phase of the alloy.

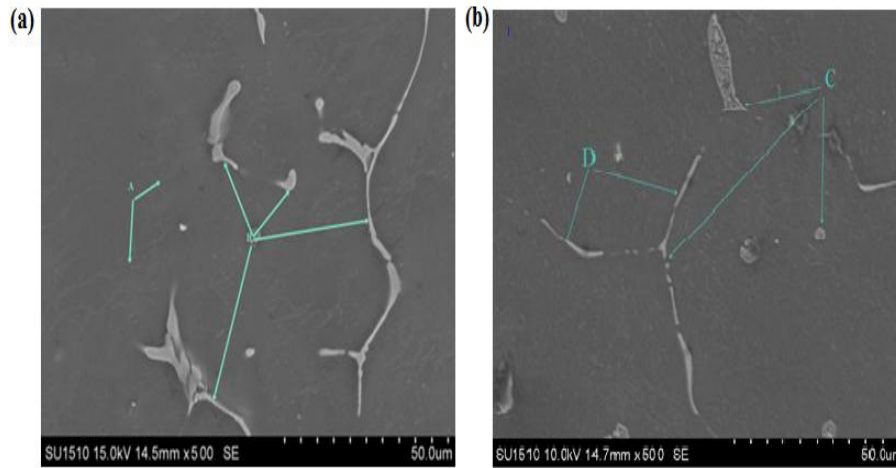


Figure 3. SEM micrographs (a) base alloy, (b) base alloy + 1.5 wt. % Sm

SEM images also showed that the microstructure refined as addition of Sm and the grain boundaries become extremely distinct. In addition, the eutectic phase showed as a needle and granular morphology distributed randomly, Figure 3-b, and the granular compounds distributed into the α -Mg matrix and the grain boundaries can restrain deformation of the matrix (Xu and Han 2012). Another researchers reported that the SEM micrographs of the as-cast Mg-6Zn-0.4Zr alloy and Mg- 6Zn-4Sm-0.4Zr alloy consists of black α -Mg, grey eutectic phases distributed along grain boundaries and a few isolated particles inside the grains (Zhang, Huang et al. 2017).

Five phases are identified via XRD technique, which are α -Mg, Mg₁₂Nd, Zn₂Zr₃, MgZn₂ and ternary phase (Mg₃Zn₆Sm) as shown in Figure 4. MgZn₂ and Zn₂Zr₃ phases which have small amounts and dissolved inside the Mg matrix of base alloy, Zn₂Zr₃ has a tetragonal crystal structure and (a 3.3030 Å - b 3.3030 Å - c 11.2600 Å) as crystallographic parameters, while MgZn₂ has a hexagonal crystal structure and crystallographic parameters (a 25.7760 Å - b 25.7760 Å - c 8.7620 Å).

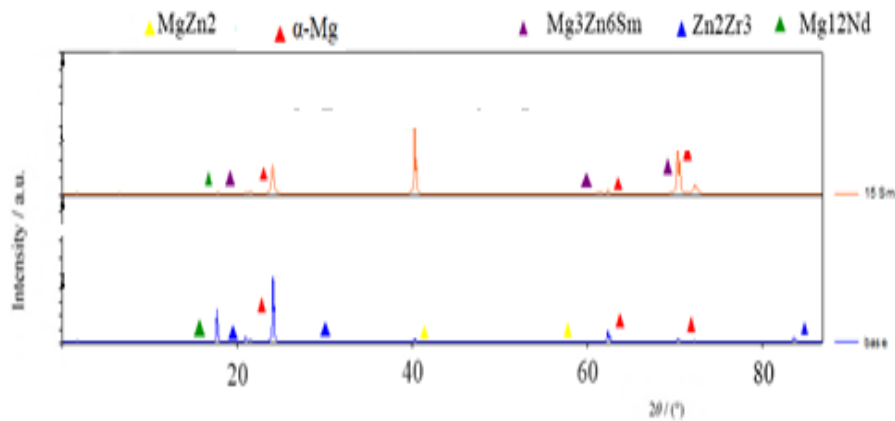


Figure 4. XRD patterns of Mg-Nd-Gd-Zn-Zr -1.5 wt.% Sm alloys

3.3 Mechanical Properties

As shown in Table 1, addition of Sm modified the values of Ultimate Tensile Strength(UTS).The UTS for base alloy with the value of 109 MPa. However, with 1.5 wt. %Sm, the UTS increased sharply to 127.56 MPa. The results of this study are in good agreement with that obtained from the relationship between tensile properties and grain size of the Mg-6Zn-0.4Zr magnesium alloy (Zhang, Huang et al. 2017). The improvement in tensile properties is attributed mainly to refined grains and the formation of Al₂Sm intermetallic compounds(Chen, Jin et al. 2015).

Table 1. Tensile strength results of base alloy and base alloys treated with Sm

Sm (wt.%)	UTS (MPa)	YS (MPa)	EI (%)
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0	109	82.09	5.934
1.5	127.56	97.6	8.176

The YS increased sharply with the addition of 1.5 wt. % Sm, obtaining 97.6 MPa compared to 82.09 MPa for the base alloy. The yield strength is dependent on RE content and increases with more RE, because of secondary and ternary phases that appears when RE add to magnesium alloy, which is due to decreasing of grain size. Sm rare earth element has a high solid solubility into the magnesium alloy, therefore, the solid solution strengthening has an important role in improving the mechanical properties of the Mg alloys treated with Sm (Li, Liu et al. 2014). Addition of Sm led to sharply increases the elongation (El) of alloy, which were 5.934 and 8.176 % for 0 and 1.5 wt.% Sm respectively, which attributed to the effect of the formed intermetallic phases. Previous studies showed that the elongation of Mg alloys were increased as addition of Sm (ZHANG, HUANG et al. 2017), the intermetallic phases can provide very effective hindrances to grain boundary sliding and dislocation motion during deformation process, which leads to improved tensile properties (Sun, Hu et al. 2015 & Hussain et al, 2020).

The hardness value increased with addition of Sm, which obtained 60.4 Hv compared with 48.33 Hv for the untreated alloy, the improvement in hardness attributed to the presence of comparatively harder (Mg-Zn-RE) formed phases, as well as increasing its volume fraction. Moreover, the reduction in grain size (decremented by about 160 μm) as addition of Sm can also led to increase hardness value. It has been found that the addition of Sm to Mg alloys leads to increase hardness values with increasing Sm content due to the increase of volume fraction of the solute-rich precipitates (Xia, Sun et al. 2016). Moreover, Z, Y., et al. research concluded that the addition of Sm had a magnificent influence on the tensile properties and significant improvements of mechanical properties of magnesium alloy (ZHANG, HUANG et al. 2017).

4. Conclusions

The effects of added Sm on the solidification characteristics, microstructure, and mechanical properties of Mg–2.8 wt. % Nd–1.5 wt. % Gd–0.5 wt. % Zn–0.5 wt. % Zr magnesium alloy were investigated. The results are summarized as follows:

- (i) The computer-aided cooling curve thermal analysis and microstructure results showed that Sm rare earth improve solidification time and grain refinement, where the grain size of treated alloy decreased by 160 μm as decrease the solidification time by 47 s.
- (ii) SEM results showed that Sm combined with the initial phase of the intermetallic base alloy, which crystallized along the grain boundaries, and $\text{Mg}_{41}\text{Sm}_5$, $\text{Mg}_3\text{Zn}_6\text{Sm}$ and $(\text{Mg}, \text{Zn})_3\text{Sm}$ were obtained as new phases.
- (iii) The mechanical properties improved significantly as addition of Sm, were UTS and YS increased by 14.6 and 15.8 % respectively as Sm content reached 1.5 wt. %, furthermore, the hardness value improved greatly by 19.9 % as well.

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